



Selenium Nanoparticles-loaded Chitosan Microspheres as a Dietary Selenium Source in Rabbits: Impact on Meat Selenium Content and Oxidative Stability

S Fortatos¹, E Giamouri¹, AC Pappas¹, SN Yannopoulos² and G Papadomichelakis^{1*}

¹Laboratory of Nutritional Physiology and Feeding, Department of Animal Science, School of Animal Biosciences, Agricultural University of Athens, Greece

²Foundation for Research and Technology Hellas – Institute of Chemical Engineering Sciences (FORTH/ICE-HT), P.O. Box 1414, GR-26504, Rio-Patras, Greece

*Corresponding Author: G Papadomichelakis, Laboratory of Nutritional Physiology and Feeding, Department of Animal Science, School of Animal Biosciences, Agricultural University of Athens, Greece.

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Abstract

Dietary selenium (Se) supplementation is a viable strategy to enhance the antioxidant defense. The commonly-used Se sources (sodium selenite and selenium-yeast) have a narrow margin between beneficial and toxic effects. Se nanoparticles stabilized in chitosan microspheres (CS-SeNP) are well established for their low toxicity, but their bioavailability and antioxidant potential has not been extensively investigated in livestock feeding. Our objectives were: a) to synthesize and characterize the properties of CS-SeNP and b) to compare the effects of CS-SeNP as dietary selenium source with those of sodium selenite and selenium-yeast on meat selenium concentration and oxidative stability in growing rabbits. The CS-SeNP were synthesized using a chemical reducing method and were characterized by dynamic light scattering, X-ray diffraction and X-ray photoelectron spectroscopy. Four experimental diets were offered to 96 rabbits; one control (C) with no added Se, and 3 diets supplemented with 0.4 mg Se/kg either from sodium selenite + selenium-yeast (1:1 ratio; T1), selenium-yeast and CS-SeNPs (1:1 ratio; T2) or CS-SeNP alone (T3). Feed intake, weight gain and feed conversion ratio were monitored throughout the trial. At the end of the trial, 12 rabbits per diet were sacrificed and meat samples were collected. Meat fatty acid composition was determined by gas chromatography. Meat Se content and oxidative stability were determined by hydride (vapor) generation atomic absorption spectroscopy and iron-induced lipid oxidation, respectively. Spherical monodispersed CS-SeNPs of 80.5 ± 20 nm average diameter were obtained. The CS-SeNP were exclusively composed of elemental Se and were totally encapsulated in chitosan, as indicated by the X-ray diffraction and X-ray photoelectron spectroscopy surveys, respectively. Growth performance was not affected by the dietary selenium addition and differences were observed between the dietary selenium sources. Meat selenium content and oxidative stability was similar in T1, T2 and T3 rabbits, but significantly higher ($P < 0.05$) when compared to C rabbits. In conclusion, the selenium from CS-SeNP enriches meat with selenium and enhances meat oxidative stability in a manner similar to the commonly-used inorganic and organic forms. Given their well-established low toxicity, CS-SeNP have a very good potential as dietary Se source and should be further studied.

Keywords: Meat; Oxidative Stability; Selenium; Selenium Nanoparticles-loaded Chitosan Microspheres; Selenium-yeast; Sodium Selenite

Abbreviations

CS: Chitosan; DLS: Dynamic Light Scattering; CS-SeNPs: Selenium Nanoparticles-Loaded Chitosan Microspheres; Se: Selenium; SeNPs: Selenium Nanoparticles; XRD: X-ray Diffraction; XRS: X-ray Photoelectron Spectroscopy

Introduction

Selenium (Se) is an integral part of at least 25 selenoproteins, some of which are involved in cellular antioxidant defense and redox regulation [1,2]. Animals can readily incorporate dietary Se into edible tissues and produce Se-enriched meat [3] with improved oxidative stability. Indeed, studies in broilers [4] and rabbits [5-7] have showed that feeding supplemental Se may significantly retard oxidation in liver and meat, in addition to tissue enrichment with Se.

To meet the daily requirement of Se and achieve balanced antioxidant status, extra dietary Se supplementation is necessary in animals reared under commercial conditions. The inorganic sodium selenite and organic Se yeast are commonly-used Se sources in the feed industry [8,9], with organic Se being more bioavailable [10,11] and less toxic [12]. Nevertheless, the margin between beneficial and toxic effects in both forms is considered narrow thus, resulting in certain regulations, which limit the maximum addition in all farm animals so that total dietary Se level will not exceed 0.5 mg/kg [13]. However, there is widespread concern that these recommendations may not be sufficient to prevent selenium deficiency and therefore, there is continued research into alternative selenium sources [14] with greater bioavailability and/or lower toxicity that may be used at higher dietary levels.

With the great progress achieved in the nanotechnology field, selenium nanoparticles (SeNP) have gained considerable attention, because of their unique features such as a high surface activity, high catalytic efficiency, large surface area, strong adsorbing ability, but most importantly low toxicity [15-17]. Acute toxicity tests have indicated that SeNP were much safer than sodium selenite [18] and organic selenium [19], and moreover, exhibited strong antioxidant properties [18,19]. However, SeNP usually enlarge, aggregate and finally transform into a gray/black analog that is thermodynamically stable, but biologically inert [18]. Several compounds can be used to stabilize SeNP including but not limited to polysaccharides like chitosan (CS) and proteins such as bovine serum albumin (BSA) [18]. Generally, BSA as a hydro-soluble protein, could be easily dissolved in the digestive tract of animals thereby releasing SeNPs faster. In contrast, CS might retard the release of SeNP due to the deficiency of the enzymes to deal with it in some species of animals and human beings, but is considered superior to BSA in terms of biosafety [18]. In contrast to the expected slower Se release, several studies in mice have showed that selenium nanoparticles-loaded chitosan microspheres (CS-SeNP) is a safe, quite stable form with an acceptable bioavailability and high antioxidant potential [18,19].

The up to date work in rabbits includes limited studies, which mainly used SeNP stabilized with BSA and other factors with quite good effects on semen quality [20] and growth performance [21] in heat-stressed rabbits. However, none of these studies performed a direct comparison of SeNP with the commonly-used inorganic and organic Se forms. Moreover, the potential of SeNP regarding bioavailability and antioxidant potential was not investigated. Therefore, the present study sought to a) synthesize and characterize the physico-chemical properties of CS-SeNP and b) compare the efficacy of CS-SeNP with that of sodium selenite (SS) and Se-yeast (SY) in rabbits, at the recommended by the EU guidelines dietary levels. The comparison between dietary Se forms herein focused mainly on meat Se content and oxidative stability to gain an indirect insight in bioavailability and antioxidant potential.

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Materials and Methods

Synthesis and characterization of selenium nanoparticles-loaded chitosan microspheres

Selenium nanoparticles-loaded chitosan microspheres (CS-SeNP) were synthesized by reducing sodium selenite in the presence of chitosan, which adheres to Se atoms and control the size of their aggregation according to the method of Bai, *et al.* [18] with some modifications. Briefly, 1.0 g of chitosan (CS, molecular weight 50-190 kDa, 90% deacetylated chitin; Sigma-Aldrich) and 1.6 g of ascorbic acid were dissolved in 100 ml of 1% (w/w) acetic acid. Subsequently, 10 ml of an aqueous solution containing 0.4 g of sodium selenite was added dropwise into the CS/ascorbic acid solution and stirred continuously (600-800 rpm) to obtain a red SeNP-M colloid. The colloid was dialyzed against 1% (w/w) acetic acid (3.5 kDa MWCO dialysis bags) for 6 hours to remove the excess ascorbic acid and other by-products. After that, the colloid was well mixed with a clean CS solution (3g CS in 100 ml of aqueous 1%

acetic acid) to achieve final concentrations of 0.09% (w/w) Se and 2.0% (w/w) CS. The final colloid was lyophilized and the cake was milled through a 0.5-mm sieve (Cyclotec, 1093 sample mill; Tecator, Höganäs, Sweden) to obtain a bright red free flowing powder, which was stored at room temperature until use.

The hydrodynamic sizes of the CS-SeNP were measured using dynamic light scattering (DLS) on a Mastersizer particle size and zeta potential analyzer (Malvern Instruments, Malvern, UK), with the average size being 80.5 ± 20 nm. The X-ray diffraction (XRD) was used to study the structure of the CS-SeNP and was performed in the range $5^\circ \leq 2\theta \leq 90^\circ$ with the aid of a D8 Advance diffractometer (Bruker Co, Billerica, MA, US), which utilizes the CuK α (1.5406 Å) radiation source operated at 40 kV and 40 mA. The X-ray photoelectron spectroscopy (XPS) was used to study the composition of the CS-SeNP and the experiments were carried out in an ultra-high vacuum system using the unmonochromatized AlK α line (1253.6 eV). Two analyzer pass energies were used, one at 20 eV (resulting in a full width at half maximum of 0.9 eV for the Ag3d5/2 peak) and another at 40 eV. The XPS core level spectra were analyzed using a fitting routine, which can decompose each spectrum into individual mixed Gaussian-Lorentzian peaks after a Shirley background subtraction. Errors in our quantitative data were found in the range of ~10%, (peak areas) while the accuracy for binding energies (BE) was approximately 0.1 eV. The analyzed area on the sample surface was a rectangle with dimensions 7.0×15 mm².

Animals and diets

Ninety-six healthy 35-day-old weaned New Zealand \times California male animals were purchased from a breeding farm for meat rabbits. Upon arrival at the experimental facilities, they were randomly allocated into four groups, namely control (C), T1, T2 and T3, of 24 rabbits each. They were kept indoors under natural environmental conditions in individual wire mesh cages equipped with metal troughs and automatic nipple drinkers. The rabbits of control group were fed a basal pelleted diet (BD), without any Se supplementation; the only Se present was that of feed ingredients (Table 1). The T1 rabbits were fed the basal diet supplemented with 0.2 mg Se/kg from sodium selenite (SS) and 0.2 mg Se/kg from a yeast source (SeY; Sel-Plex®, Alltech Inc, Nicholasville, KY, US). The T2 rabbits were fed the basal diet supplemented with 0.2 mg Se/kg from SeY and 0.2 mg Se/kg from CS-SeNP, which were synthesized in our laboratory as described above. Finally, the T3 rabbits were

fed the basal diet supplemented with 0.4 mg Se/kg from CS-SeNP only. These combinations were selected so as to follow the EU recommendations [13] according to which: a) dietary organic Se addition is limited to 0.2 mg/kg, b) the total Se contained in the diet (supplemental and natural occurring) should not exceed 0.5 mg/kg. Thus, the experimental design resulted in the addition of 0.2 mg Se from SS and 0.2 mg Se from SY in T1 and 0.2 mg Se from SY and 0.2 mg Se from CS-SeNP in T2 diets. The T3 diet containing only CS-SeNP was designed for comparison (vs. T1 and T2 diets) reasons. The BD was formulated according to the recommendations for growing rabbits [22]. The ingredients and chemical composition of the diets are shown in table 1.

Experimental procedures

Handling and care of the experimental animals conformed to the guidelines of the Department of Animal Science. The experimental protocol was approved by the AUA Bioethics Committee (no. 13/16-03-2021). During the experiment, body weight and feed intake were recorded weekly. At the end of the experiment (77 days of age) rabbits were sacrificed by electro stunning and exsanguination, eviscerated and after a 24 h chilling period at 4°C, carcasses were weighed and dressing percentage was determined. Subsequently, samples from the *Longissimus lumborum* (LL) muscle were collected. In detail, from each carcass, the right part of the LL muscle was excised, vacuum packed and stored at -20°C until analyzed for FA. A smaller portion (ca. 2-3 g) from the right part of the LL muscle was used for the determination of Se concentration. Simultaneously, the left part of the muscle was removed, vacuum packed and stored at -20°C, so as to determine lipid oxidation.

Determination of selenium in CS-SeNP, diets and meat

Selenium concentration in feed, CS-SeNP and LL samples was determined using an Agilent 240FS AA atomic absorption spectrometer fitted with VGA77 Vapor generation accessory (Agilent, Santa Clara, USA) based on the method described by Pappas, *et al.* [23] with minor modifications. In brief, samples (0.50 g) were soaked in 4 ml of concentrated HNO₃ (65% w/v, Suprapur; Merck, Germany). The samples were left for 30 minutes for pre-digestion. Then, another 6 ml of concentrated HNO₃ were added and then were heated in a microwave accelerated digestion system (CEM, Mars X-Press, Matthews, NC, USA) according to the following program: power was ramped during 20 min from 100 to 1200W and

Ingredient	Basal diet	
Dehydrated alfalfa	263.0	
Barley grain	266.0	
Sunflower meal, 30%	170.0	
Wheat bran	150.0	
Sugar beet pulp	120.0	
Vegetable oil	14.0	
L-Lysine HCl, 80%	2.5	
L-Threonine, 99%	1.5	
DL-Methionine, 99%	1.0	
Calcium carbonate	1.0	
Mono-calcium phosphate	1.0	
Sodium chloride	2.0	
Thyme extract (Addarome)	0.5	
Ultrafed ^{®1}	4.0	
Premix ²	3.5	
Calculated chemical composition		
Dry matter	904.0	
Organic matter	931.0	
Crude protein	157.0	
Ether extract	3.4	
NDF	338.0	
ADF	190.0	
Lysine	8.0	
Methionine + Cystine	6.0	
Threonine	6.7	
Calcium	8.0	
Phosphorus	5.5	
Digestible energy, MJ/kg	10.2	
	Se level (mg/kg as-fed)	
Diet	Added³	Determined⁴
C	-	0.093 ± 0.018
T1	0.40	0.508 ± 0.055
T2	0.40	0.516 ± 0.060
T3	0.40	0.478 ± 0.058

Table 1: Ingredient (g/kg as-fedbasis) and chemical composition (g/kg dry matter) of the basal diet, and selenium (Se) level of the experimental diets.

¹ Contained >95% palygorskite [(Mg,Al)₂Si₄O₁₀(OH)·4(H₂O)] as agglomerant (binder).

² Premix provided per kg diet: vitamin A, 12,000 IU; vitamin D3, 1,200 UI; vitamin E, 50 UI; vitamin K3, 2 mg; vitamin B1, 2.5 mg; vitamin B2, 4 mg; vitamin B6, 2 mg; vitamin B12, 0.02 mg; pantothenic acid, 12.5 mg; nicotinic acid, 40 mg; folic acid, 1 mg; biotine, 0.15 mg; choline chloride, 100 mg; I, 1 mg; Mn, 80 mg; Cu, 15 mg; Zn, 80 mg; Fe, 40 mg; Co, 0.5 mg; 300 mg Cycostat (60 mg robenidine/kg). It did not contain any Se source (organic or inorganic).

³ Se was added as: a) sodium selenite (0.2 mg/kg) and Se-yeast (0.2 mg/kg) in T1 diet, b) Se-yeast (0.2 mg/kg) and CS-SeNP (selenium nanoparticles-loaded chitosan microspheres 0.2 mg/kg) in T2 diet and c) CS-SeNP (0.4 mg/kg) in T3 diet; Se-yeast was in the form of Sel-Plex[®] (Alltech Inc., Nicholasville, KY, USA); no Se was added to the control (C) diet.

⁴ Average of 4 samples per diet ± standard deviation.

held for 15 min. The temperature reached a maximum of 200°C followed by a cool-down cycle for 15 min. Losses of volatile element compounds did not occur as the tubes were sealed during heating. The samples were then filtered with disposable syringe filters 0.20µm/15mm (Chromafil, Macherey-Nagel, Germany). Se^{VI} reduced to Se^{IV} by warming with 6M concentrated HCL (Merck, Germany) before analysis in the AA atomic absorption system. Calibration standard solutions were prepared from high purity standards. Reductant agent NaBH₄ 0.6% w/v (Sigma- Aldrich, USA) and NaOH 0.5% w/v (Fisher Scientific, UK) and 10 M HCl (Merck, Germany) was used for vapor generation. To assess the accuracy of the process, two standard reference materials were used, namely RM8414 (Bovine muscle powder; National Research Council, Canada) and RM1577c (Bovine liver; National Institute of Standards and Technology, USA).

Determination of fatty acid profile

Any external fat and connective tissue were dissected out of the muscle samples, which were then blended in a domestic food processor until smooth. Blending was performed in short bursts to ensure the homogeneous distribution of intramuscular fat in the sample. The FA of diets and intramuscular fat were extracted and methylated directly [24]. Duplicate 1 (± 0.05) g samples were hydrolyzed for 1.5 h at 55°C in 1 N potassium hydroxide in methanol, containing a known amount (approximately 0.5 mg) of tridecanoic acid (C13:0) as internal standard. The potassium hydroxide was then neutralized, and the free FAs were methylated by sulphuric acid catalysis (24N H₂SO₄) for 1.5 h at 55°C. Hexane (3 ml) was added to the reaction tube, which was vortex-mixed and centrifuged at 1100 g. The supernatant hexane layer containing the FA methyl esters was kept at -20°C, until analyzed by gas chromatography. A temperature-programmed run was followed on a Perkin Elmer Autosystem XL gas chromatograph equipped with a 30 m×0.25 mm×0.25 µm internal diameter HP-Innowax capillary column (Agilent Technologies, J&W GC columns, Santa Clara, CA, USA) and a flame ionization detector (FID). The column temperature was programmed for 1 min at 140°C, raised by 2.50 C/min to 200°C, then to 230°C by 10 C/min and held for 1 min, and finally to 240°C by 4°C/min and held for 10 min. Helium was the carrier gas at a constant pressure of 18 psi and the temperature of both the injector and FID was set at 250°C. Fatty acids were identified by comparison with standards purchased from Sigma-Aldrich Co. (FAME 37 Component; Sigma-Aldrich Co. Supelco, IL, USA) and quantification was achieved using the internal standard (13:0) added prior to hydro-

lysis. Total weights of FA (mg/100g) in diets were calculated as the sum of areas for all FA peaks compared to area for 0.5 mg internal standard. Individual FA were expressed as % by weight of total FA.

Determination of iron-induced lipid oxidation

Lipid oxidation was determined in raw and cooked meat. The malondialdehyde (MDA) formed during iron-induced lipid oxidation [25,26] was assessed. Briefly, the LL samples were thawed and homogenized in a food processor until smooth. The paste was immediately used to measure lipid oxidation in raw meat. Small patties were formed (5 cm in diameter, 1 cm thick) and cooked in a domestic microwave oven at 1000 W for 30 sec until an internal temperature of 70°C was reached [27]. Four 1.00 (± 0.05) g subsamples from the raw or cooked meat were weighed into 50-ml centrifuge tubes and 1.5 ml of a solution containing 1.138 mM ferrous sulphate and 0.368 mM ascorbic acid was added to three of the sub-samples, which were incubated at 37°C for either 50, 150 or 300 min. Following incubation, all three iron-induced subsamples along with the 4th non-induced subsample were immediately homogenized (Polytron homogenizer, PCU, Littau/Lucerne, Switzerland) in the presence of 8 ml of aqueous trichloroacetic acid (50 g/l) and 5 ml of butylated hydroxytoluene (BHT) in hexane (8 g/l), and the mixture was centrifuged. The top layer was discarded and the bottom layer was filtered (Macherey-Nagel no. MN 1640W, Düren, Germany). A 2.5-ml aliquot of the filtrate was mixed with 1.5 ml of aqueous 2-thiobarbituric acid (8 g/l) and further incubated at 70°C for 30 min. Subsequently, the mixture was cooled under tap water and submitted to spectrophotometry (Helios α, Thermo spectronic, Cambridge, UK) at 532 nm. The concentration of MDA (mg/kg) in samples was calculated by referring to slope and intercept data of the standard calibration curve (0 – 0.96 µM) prepared using 1,1,3,3-tetraethoxypropane (TEP), previously hydrolyzed in 0.1N HCl.

Statistical analysis

Data were analyzed using the SPSS statistical package (version 17.0) and are presented as means ± standard error (SEM). Feed intake, body weight gain, feed conversion ratio, cold carcass weight, dressing percentage and Se content, FA profile in the LL muscle were analyzed by a one-way analysis of variance (ANOVA) with diet as fixed effect. The MDA values were analyzed using a repeated measures ANOVA with diet and time as fixed effects. Post-hoc tests were conducted based on Tukey's criterion. Statistical significance was set at P < 0.05 for all tests.

Results and Discussion

Characterization of the Se nanoparticles-loaded chitosan microspheres

The DLS analysis showed that monodispersed spherical CS-SeNP with an average diameter of 80.5 ± 20 nm (mean \pm s.d.) were produced. The XRD pattern showed a totally amorphous structure of the Se in the CS-SeNP, since no traces of Bragg reflections related to crystalline Se were observed (Figure 1). The XPS survey scans revealed the presence of C, O, and N atoms on the sample surface (Figure 2a). The C1s peak was deconvoluted into three components as shown in figure 2b. The band located at 284.7 eV, could be assigned to C-C or adventitious carbon, which confirmed the presence of the aliphatic group. The band at 286.3 eV was associated with C-N and C-O groups, and the band at 288 eV was attributed to C=O or O-C-O species. In addition, the N1s peak centered at 399.3 eV (Figure 2c), could be assigned to $-NH_2$ and $-NH$ groups. The Se3d peak centered at 55.3 eV (Figure 2d) confirming that the valence state of Se in CS-SeNP was zero (Se^0). Based on the XPS peak areas, the relative atomic concentration on the surface of CS-SeNP was calculated and presented in table 2. High concentrations of C, O, and N atoms (67.97, 28.67 and 3.12%, respectively) corresponding to CS along with traces of Se (0.25%) were found on the surface of SeNP-M. The average Se content was determined at 20000 mg/kg CS-SeNP.

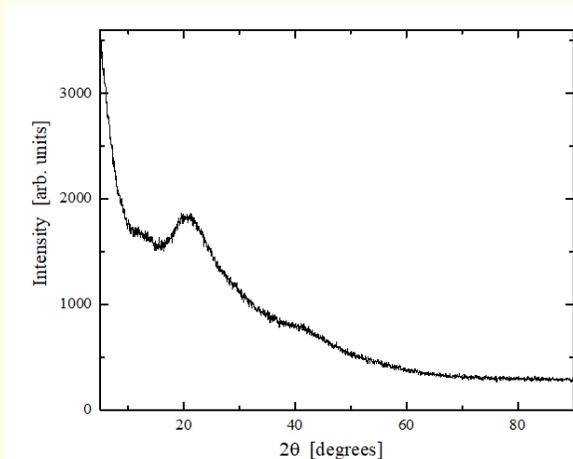


Figure 1: X-ray diffraction (XRD) patterns of the selenium nanoparticles-loaded chitosan microspheres (CS-SeNP).

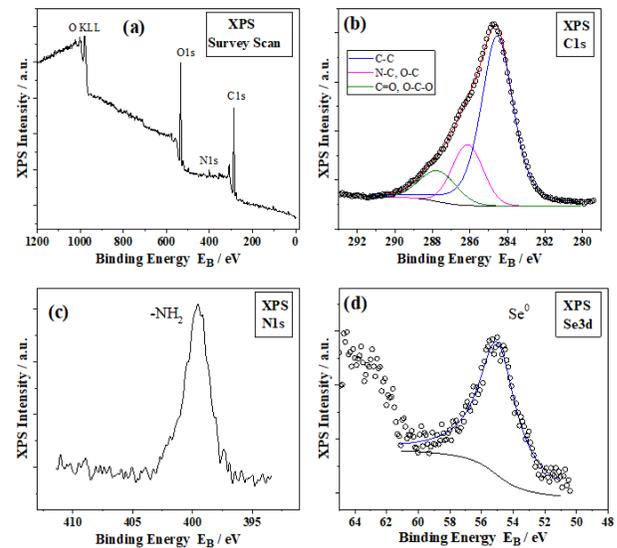


Figure 2: (a) Survey X-ray photoelectron spectroscopy (XPS) spectrum of the selenium nanoparticles-loaded chitosan microspheres (CS-SeNP). (b) Deconvolution of the C1s XPS spectrum into three components. (c) N1s XPS spectrum. (d) Se3d XPS spectrum. Experimental data appear as open circles in (b) and (d), while solid lines are fitting results.

Since CS-SeNP were synthesized in the laboratory using some modifications on a previously published method [18], our first concern was to confirm their characteristics before supplementing the diets. The size distribution analysis conducted using DLS calculated the average diameters of CS-SeNP to be very close (80.5 ± 20.0 nm) to the 95 nm previously reported [18]. The elemental nature of Se in the CS-SeNP was also confirmed by XPS technique in agreement with Bai, *et al.* [18]. However, the elemental Se in the CS-SeNP prepared herein was amorphous, in contrast to the crystalline (trigonal phase) feature reported by Bai, *et al.* [18]. It is not unlikely that differences in the manufacturing conditions between our study and that of Bai, *et al.* [18] may have affected the nature of Se in the SeNP. Zhang, *et al.* [28] observed that amorphous elemental Se was present in SeNP encapsulated in high molecular weight CS (200 kDa), whereas crystalline elemental Se was obtained with low molecular weight CS (3 kDa). The crystalline nature of Se when

using a very low molecular weight CS (3 kDa) was also reported by Bai, *et al.* [18]. Herein, we used a moderate to high molecular weight CS (50-190 kDa), which may have resulted in amorphous Se. In general, CS-SeNP with desired properties could be easily prepared in the laboratory and produced in large quantities for the needs of the experiment quite fast. This indicates that CS-SeNP synthesis could possibly be up-scaled to industrial production, as the method adopted here was economic and environment-friendly [18].

Element	Binding energy (eV)	Concentration (%)	Assignment
O1s	532.53	28.67 ± 0.05	C-O
C1s	284.75	67.97 ± 0.06	C-C, C-N, O-C
N1s	399.55	3.12 ± 0.04	-NH ₂ , -NH
Se3d	55.28	0.25 ± 0.01	Se (0)

Table 2: Percent relative atomic concentration of elements on the surface of the selenium nanoparticles-loaded chitosan microspheres (CS-SeNP) as determined using the peak areas of the X-ray photoelectron spectroscopy (XPS) spectra.

Growth performance and carcass traits

The diet did not have any effect on feed intake, growth rate and feed conversion ratio in growing rabbits. Also, no effect was observed on cold carcass weight and dressing percentage (Table 3). Feed intake, weight gain, feed conversion ratio and carcass traits were not affected by the combination of SS with SY (T1 diet), of SY with CS-SeNP (T2) or the CS-SeNP alone (T3) in comparison with the unsupplemented C diet. Abdel-Wareth, *et al.* [20] reported significantly improved body weight gain in male reproductive rabbits fed diets with nano-elemental Se encapsulated in bovine serum albumin (BSA) due to the improved nutrient digestibility. Sheiha, *et al.* [21] also observed that rabbits fed with bio- or chemically synthesized Se nanoparticles had better growth performance and dressing percentage when compared to a control diet, with biological nano-Se being more efficient. However, in both studies [20,21] the experiments were carried out under severe heat-stress conditions, which was not the case in the present work; temperature and humidity ranged from 23-26°C, and 50-70%, respectively. Also, in both studies the Se content in the control diets was not clear and no comparison of nano-Se with inorganic or organic Se forms was con-

ducted. Earlier works with dietary inorganic or organic Se in rabbits suggested that positive effects of supplemental Se on growth can be found only when the basal diet is Se-deficient [5,7] or when rabbits are kept under severe environmental conditions [6]. It can be assumed that the Se contained in the basal (control) diet herein satisfied the minimum requirements in growing rabbits [22]; thus, supplemental Se in the form of SS, SY or CS-SeNP was not expected to affect the performance of rabbits.

	Diet ¹				SEM ²	P-value
	C	T1	T2	T3		
Initial BW, g (35 d)	1005	1000	1013	1012	34.3	0.793
Final BW, g (77 d)	2952	3010	2943	2975	100.1	0.916
ADFI, g/d	157.9	161.2	160.6	156.3	6.48	0.863
ADWG, g/d	47.8	49.6	47.8	48.0	1.83	0.740
FCR, g/g	3.31	3.27	3.35	3.26	0.071	0.537
Cold carcass weight, g	1763	1813	1774	1784	71.3	0.906
Dressing percentage, %	62.7	63.9	63.8	63.6	0.65	0.265

Table 3: Effects of diet on body weight (BW), average daily feed intake (ADFI), average daily weight gain (ADWG), feed conversion ratio (FCR) and slaughter traits (n = 24 rabbits/diet).

¹ Se source: a) sodium selenite (0.2 mg/kg) and Se-yeast (0.2 mg/kg) in T1 diet, b) Se-yeast (0.2 mg/kg) and CS-SeNP (selenium nanoparticles-loaded chitosan microspheres, 0.2 mg/kg) in T2 diet and c) CS-SeNP (0.4 mg/kg) in T3 diet; Se-yeast was in the form of Sel-Plex® (Alltech Inc., Nicholasville, KY, USA); no Se was added to the control (C) diet.

² SEM= Standard error of means.

Meat selenium content, fatty acid composition and oxidative stability

The T1, T2 and T3 diets supplemented with Se increased significantly (P<0.05) muscle Se content by 196, 182 και 174%, respectively, in comparison with the control diet. No differences in the muscle Se content were found between T1, T2 and T3 diets (Figure 3). Based on our experimental design, it is assumed that SS, SY and CS-SeNP enriched meat Se to a similar extent.

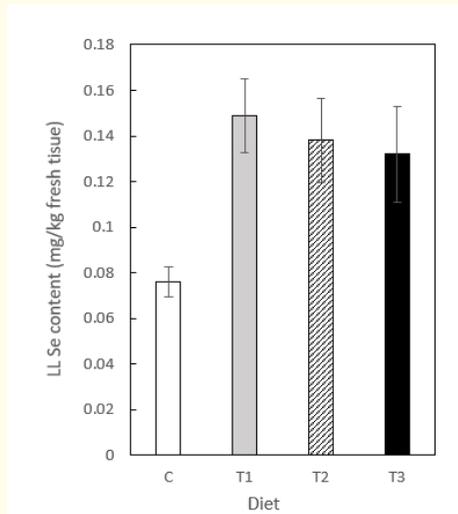


Figure 3: Effect of diet on Se content (mg/kg wet tissue) of *Longissimus lumborum* (LL) muscle in 77 day-old rabbits (n = 12 rabbits/diet).

One key aspect of dietary Se supplementation in rabbits is enrichment of meat with Se [6,7,29,30] with health promoting effects for humans. The organic forms, such as SY, are considered more effective in increasing muscle Se concentrations [11,30-32], because

most Se is in the form of seleno-methionine, which has higher bioavailability [33]. The effect of the dietary nano-Se supplementation on meat Se content and therefore, their bioavailability has not been studied in rabbits. Zhang, *et al.* [28] reported a Se bioavailability from chemically synthesized CS-SeNP comparable to that of SS in mice. Bai, *et al.* [18,34] also reported that SeNP in CS-SeNP contributed to the Se retention in a dose-dependent manner in mice, but Se retention of CS-SeNPs was weaker than that of SS. Generally, the SeNP are envisaged widely, particularly in biomedicine, due to their high bioavailability [35]. However, this does not appear to be the case for all SeNP preparations. Our results showed that Se from CS-SeNP accumulated efficiently to muscle tissue, as SS and SY did, despite the fact that CS is considered to delay absorption of Se [18]. Most likely, CS as acid-soluble polysaccharide was dissolved to some extent by the stomach acids or by the free radicals formed by the intestinal cells activity during absorption of nutrients. Such mechanisms have been investigated and proposed in previous studies [28,34].

No major differences in the meat FA composition were detected between diets. Total saturated FA (SFA) were higher (P< 0.05) in T3 compared to T1 and T2 fed rabbits only. The 20:5n-3 was lower (P< 0.05) in T3 compared to T1 fed rabbits only. However, these differences were not enough to substantially modify the polyunsaturated FA (PUFA) or the n-3 content (Table 4) in the intramuscular fat.

	Diet ¹				SEM	P-value
	C	T1	T2	T3		
Total FA	943	932	958	937	30.2	0.991
14:0	1.35	1.38	1.31	1.49	0.051	0.635
14:1	0.11	0.17	0.17	0.12	0.013	0.234
15:0	0.34	0.34	0.32	0.33	0.005	0.216
16:0	24.52	23.73	23.73	24.97	0.221	0.116
16:1n-9	0.33ab	0.36a	0.32b	0.30b	0.005	0.001
16:1n-7	2.46	2.97	2.92	2.55	0.167	0.628
17:0	0.41a	0.39ab	0.37b	0.37b	0.006	0.015
17:1	0.21	0.23	0.20	0.20	0.006	0.200
18:0	8.35ab	8.14a	8.63b	8.82b	0.084	0.020
18:1n-9	20.74	21.05	20.61	21.26	0.279	0.851
18:1n-7	1.42	1.51	1.41	1.36	0.026	0.203
18:2n-6	18.34	18.65	18.73	18.64	0.236	0.946

18:3n-3	0.94	0.93	0.92	0.97	0.020	0.775
20:1n-9	0.26	0.25	0.25	0.27	0.007	0.610
20:2	0.78	0.81	0.80	0.73	0.016	0.315
20:3n-6	0.82	0.85	0.82	0.70	0.022	0.075
20:3n-3	0.15	0.15	0.15	0.17	0.010	0.765
20:4n-6	6.01	5.84	5.95	5.38	0.179	0.605
20:5n-3	0.20 ^{ab}	0.22 ^a	0.20 ^{ab}	0.17 ^b	0.006	0.038
22:2	0.27	0.26	0.26	0.23	0.006	0.177
22:4n-6	1.79	1.66	1.67	1.53	0.050	0.344
22:5n-3	0.75	0.74	0.72	0.64	0.025	0.400
22:6n-3	0.15	0.15	0.15	0.13	0.007	0.540
ΣSFA ³	35.16 ^{ab}	34.18 ^b	34.55 ^b	36.26 ^a	0.237	0.008
ΣMUFA ³	25.88	26.88	26.25	26.45	0.463	0.901
ΣPUFA ³	30.29	30.36	30.44	29.38	0.462	0.845
ΣPUFA/ΣSFA	0.87	0.90	0.89	0.82	0.018	0.423
Σn-3 ⁴	2.20	2.19	2.13	2.08	0.029	0.483
Σn-6 ⁴	27.05	27.10	27.25	26.33	0.430	0.885
ΣLCn-3 ⁵	1.25	1.26	1.21	1.11	0.041	0.558

Table 4: Effects of diet on total fatty acids (mg/100 g wet tissue) and fatty acid (FA) profile (% of total FA) of *Longissimus lumborum* in 77 day-old rabbits (n = 12 rabbits/diet).

¹ Se source: a) sodium selenite (0.2 mg/kg) and Se-yeast (0.2 mg/kg) in T1 diet, b) Se-yeast (0.2 mg/kg) and CS-SeNP (selenium nanoparticles-loaded chitosan microspheres, 0.2 mg/kg) in T2 diet and c) CS-SeNP (0.4 mg/kg) in T3 diet; Se-yeast was in the form of Sel-Plex® (Alltech Inc., Nicholasville, KY, USA); no Se was added to the control (C) diet.

²SEM= Standard error of means.

³ΣSFA= Sum of saturated fatty acids (12:0+14:0+15:0+16:0+17:0+18:0), ΣMUFA= sum of monounsaturated fatty acids (14:1+16:1n-7+17:1+18:1n-9+18:1n-7+20:1n-9), ΣPUFA= sum of polyunsaturated fatty acids (18:2n-6+18:3n-3+18:3n-6+20:3n-6+20:3n-3+20:4n-6+20:5n-3+22:4n-6+22:5n-3+22:6n-3).

⁴Σn-3= sum of n-3 fatty acids (18:3n-3+20:5n-3+22:5n-3+22:6n-3), Σn-6= sum of n-6 fatty acids (18:2n-6+18:3n-6+20:3n-6+20:4n-6+22:4n-6).

⁵ΣLCn-3= sum of long carbon chain (≥20C) n-3 fatty acids (20:3n-3+20:5n-3+22:5n-3+22:6n-3)

In a previous study [7], it was reported that dietary supplementation with SY improved muscle FA composition in comparison with an unsupplemented diet, likely through an *in vivo* protection of polyunsaturated FA or peroxisomal β-oxidation, which benefits the synthesis of omega-3 (n-3) FA. No such effect was observed herein. This disagreement cannot be fully explained and merits further investigation.

The MDA concentrations in the raw LL samples did not differ between diets at all the designated time points, although a trend for lower MDA concentration was observed in T1, T2 and T3 fed rabbits when compared to control after 300 min of oxidation. The picture was different in the cooked LL samples, where T1, T2 and T3 fed rabbits showed significantly (P < 0.05) lower MDA concentrations compared to the C ones after 300 min of oxidation (Figure

4). Based again on our experimental design, it is assumed that SS, SY and CS-SeNP had a similar protective role against cooking-induced oxidation.

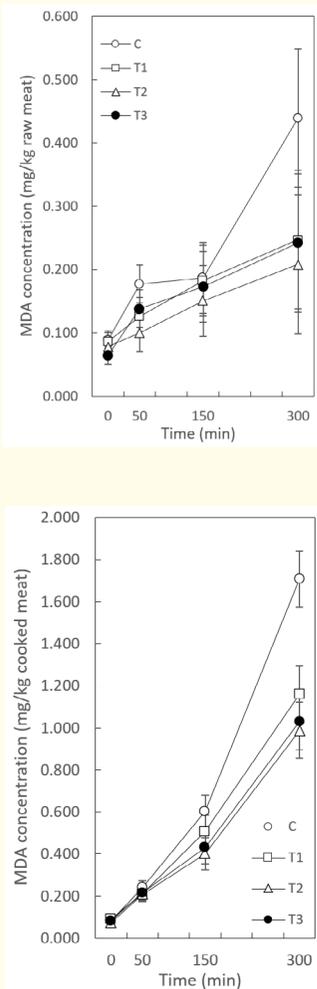


Figure 4: Effect of diet on iron-induced lipid oxidation of *Longissimus lumborum* in 77 day-old rabbits (n = 12 rabbits/diet). MDA = malondialdehyde.

The results indicate that raw meat had a very good oxidative stability, which might be explained by the thyme extract in the basal diet (Table 1). This extract contains essential oils that may have protected the meat from oxidation; therefore, the Se added to diet did not exhibit any significant effects. However, when meat was ad-

ditionally stressed by cooking in the microwaves, the protective role of Se (from SS, SY or CS-SeNP) against the control was obvious. Generally, conflicting reports on the effect of dietary Se addition on rabbit meat oxidative stability can be found in literature. In some studies, supplemental SY reduced significantly malondialdehyde (MDA) values during refrigerated storage for 6 days [6] or iron-induced lipid oxidation [7], while no such effect was observed in others [29,30] using SS and/or SY as Se source. The similar oxidative stability of meat in rabbits fed the Se supplemented diets, clearly indicated that CS-SeNP had a significant antioxidant potential, but not superior, as would be expected according to literature, to the commonly-used SS or SY sources. Our results indicate that the elemental Se from CS-SeNP was not only absorbed, it was also biologically active, in accordance Zhang, *et al.* [36], who was the first to challenge the theory that Se with zero valence state is inactive.

Conclusion

Meat Se content readily increased and meat oxidative stability was enhanced when CS-SeNP was added in the diet. These changes were similar to those induced by the dietary addition of SS or SY. The present results indicate that CS-SeNP is a source of bioavailable Se and with similar antioxidant potential compared to the commonly used Se sources. The CS-SeNP merits further investigation as feed additive, in terms of alternative (supranutritional) dietary levels.

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Conflict of Interest

Declare if any financial interest or any conflict of interest exists.

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